PCE3I001 CHEMICAL PROCESS CALCULATIONS

Module I:

Units & dimensions, Ideal gas laws, equation of state, Vapor pressure, Clausius-Clapeyron equation, Ideal and non-ideal solutions, humidity-relative saturation & percentage saturation, concept of wet & dry bulbs thermometer, use of Humidity chart.

Module II:

The chemical equation &stoichiomentry, concept of limiting & excess reactants, conversion, degree of conversion, yield etc.

Material balances &unit operations: drying, crystallization dissolution, combustion, etc. Solving material balance (steady and unsteady state processes) without and with chemical reactions, recycle, bypass,& purge calculations.

Module III:

Energy balance concepts: Heat capacity, Calculation of enthalpy changes without change of phase, Energy balance with chemical reaction, Standard heat of reaction at constant, pressure & constant volume, effect of T and P on heat of reaction, Adiabatic reaction of temperature, heat of solution & mixing.

Text Books:

- 1. Stoichiometry and Process Calculations by B Lakshmikutty and K V Narayanan, PHI.
- 2. Stoichiometry, 5th ed. by B I Bhatt and S B Thakore, McGraw-Hill.

References Books:

- 1. Elementary Principles of Chemical Processes, 3rd ed. by R M Felder and R W Rousseau, John Wiley.
- 2. Chemical Process Principles: Material and Energy Balances (Part 1), 2nd ed. by O A Hougen, K M Watson, and R A Ragatz, CBS.
- 3. Principles of Chemical Engineering Processes by N Ghasem and R Henda, CRC.
- 4. Basic Principles and Calculations in Chemical Engineering, 8th ed. by D M Himmelblau and J B Riggs, PHI.